# POARCH BAND OF CREEK INDIANS UTILITIES AUTHORITY



# 2018 ANNUAL WATER QUALITY REPORT



Draw a line matching the items on the left to the amount of water on the right.

1. Taking a shower A. 30 gallons B. 180 gallons 2. Watering the lawn C. 4-7 gallons 3. Washing the dishes 🖵 D. 1/2 gallon 4. Washing clothes 🖵 E. 39,090 gallons 5. Flushing the toilet  $\Box$ F. 62,600 gallons 6. Brushing teeth 🖵 G. 15-30 gallons 7. Drinking 🖵 H. 9.3 gallons 8. Needed to produce one ton of steel 1. 1 gallon 9. Needed to process one J. 9-20 gallons can of fruit or vegetables  $\Box$ 10. Needed to manufacture a new car and its four tires

> S-C, 6-I, 7-D, 8-F, 9-H, 10-Ε Answers: 1-G, 2-B, 3-J, 4-A





## **Understanding Your** WATER QUALITY REPORT

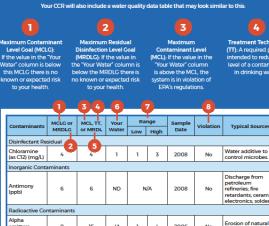
The Consumer Confidence Report (CCR) is an annual water quality report that a community water system is required by law to provide to its customers each year by July 1st. Your CCR can help you make informed choices about the water you drink.

## Your CCR Provides Need-To-Know Information



your CCR compared national standard and any violations health-based standa

## Sample Water Quality Data Table



6

8

#### WHY don't I get a CCR?



you don't pay your own water bill because you live in an apartment, ndo, or rental property or you get your water from a private ground ater well, you don't receive a CCR. Renters can contact their building anager or search online to see if your CWS publishes its CCR.

#### **Questions Or Concerns About Your CCR**



CALL EPA'S SAFE WATER CALL EPA'S SAFE WATER HOTLINE at 1-800-426-4791 if you would like to know more about your CCR, how to locate your local water company, or for more resources



HEALTHCARE PROVIDER if you are sensitive to contaminants or if you are at higher risk of infections



For more information, visit: epa.gov/ccr

# **HOW TO DRINK** MORE WATER

Your Step by Step Daily Water Drinking Plan.

## 6-8am • benefits

CUP #1AND 2 Drink 2 cold glasses of water in the morning when you wake up and

before you eat breakfast.

## 9-10am

## CUP #3

Drink one cup around morning tea time or when you start to feel peckish & before you reach for an unhealthy snack.

#### 10am - 12pm ē

#### **CUP #4**

Drink a glass of water at least 1/2 an hour before you have your lunch.

## 12pm - 2pm

**CUP #5** Drink a glass of water at least 1/2 an hour after lunch.

## 2pm - 4pm

Enjoy a cold glass of water when you start to feel that afternoon slow down.

## 4pm - 6pm 🗯

CUP #6 📍

**CUP #7** Drink this one at least 1/2 an hour before dinner.

## before bedtime

**CUP #8** Water should be served at room temp & at least 20 min before bed.



- Rehydrate your brain & wake up
- Kick start your metabolism
- by 24% before breakfast.
  - Stimulate your digestive
- system and get things moving.

## benefits

- Lubricate joints and cartilage to get your body moving.
- Get an Immunity Booster Shot and neutralize the PH in
- your body. Look younger with plumper
- hydrated skin.

## benefits

- Burn an extra 5lb of fat / year. Boost your metabolism for the next 30 - 40 minutes.
- Power up your mental
- creativity by up to 30%

## benefits

- Prevent premature aging of the major organs in your body.
- Increased production of blood and muscle cells.

## benefits

- Natural Energy Drink: Reduced fatique & increases enzyme activity in the body.
- Lowers Cholesterol production.

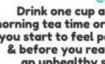
## benefits

Reduce your risk of colon cancer by 45% & bladder cancer by 50%



Eliminate toxins and inflammation causing bacteria.





## 2018 Annual Water Quality Report (Testing Performed January through December 2017)

## POARCH BAND OF CREEK INDIANS UTILITY AUTHORITY

5811 Jack Springs Road Atmore, AL 36502 Phone 251-446-1617 Fax 251-446-1624

We are pleased to present to you this year's Annual Water Quality Report. This report will tell you where your water comes from, what contaminants have been detected, and how these detection levels compare to Federal and State drinking water standards. This report is designed to inform you about the quality water and services we deliver to you every day. We want you to understand the efforts we make to continually improve the water treatment process and protect our water resources. We are committed to ensuring the quality of your water.

Water Source	Two groundwater wells producin	Two groundwater wells producing from the Miocene formation				
water Source	Purchased groundwater from Freemanville Water System					
Water Treatment	Chlorination, fluoridation and pH adjustment					
Storage Capacity	Two storage tanks with a total ca	apacity of 1.2 million gallons				
Number of Customers	Approximately 240					
	Chairman	Josh Martin				
	Vice Chairman	Ron Marshall				
	Treasurer/Secretary	Chris McGhee				
Board Members	Member	Shawn Rolin				
	Member	Patrick Strickland				
	Member	Ronald Rolin				
	Member	Charles Bray				
	Utilities Executive Director	Josh Thomas				
	Maintenance Superintendent	James Ramer				
	Customer Service Supervisor	Rebecca Black				
	Operations Manager	Shaun Livermore				
	Maintenance Foreman	Nathaniel Dortch				
Staff	Chief Operator	Bill Holmes				
Stall	Office Coordinator	Kimberly Weatherford				
	Operations Supervisor	Dempsey Rolin				
	Field Operations Technician	Trenton McGhee				
	Operations Technician	Larry Bailey Jr.				
	Maintenance Technician	Bo Slate & Trent Flowers				
	Utilities Assistants	Jerry Walker & Blake Cunningham				

## Source Water Assessment

In compliance with the Alabama Department of Environmental Management (ADEM), **Poarch Band of Creek Indians Utility Authority** has developed a Source Water Assessment that will assist in protecting our water sources. This plan provides additional information such as potential sources of contamination. It includes a Susceptibility Analysis, which classifies potential contaminants as high, moderate, or non-susceptible to contaminating the water source. The report has been completed and approved by ADEM. A copy of the report is available in our office for review.

Please help us make this effort worthwhile by protecting our source water. Carefully follow instructions on pesticides and herbicides you use for your lawn and garden, and properly dispose of household chemicals, paints and waste oil.

## **Questions?**

If you have any questions about this report or concerning your water utility, please contact **Josh Thomas**. We want our valued customers to be informed about their water utility. If you want to learn more, please attend any of our regularly scheduled meetings. They are held on the third Thursday of each month at the Utilities Office, 263 Aplin Rd, Atmore, Alabama. More information about contaminants to drinking water and potential health effects can be obtained by calling the EPA's Safe Drinking Water Hotline at (1-800-426-4791).

## **General Information**

All drinking water, including bottled drinking water, may be reasonably expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. MCL's, defined in a List of Definitions in this report, are set at very stringent levels. To understand the possible health effects described for many regulated constituents, a person would have to drink 2 liters of water every day at the MCL level for a lifetime to have a one-in-a-million chance of having the described health effect.

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and radioactive material, and it can pick up substances resulting from the presence of animals or from human activity. Contaminants that may be present in source water include:

• Microbial contaminants, such as viruses and bacteria, which may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.

• Inorganic contaminants, such as salts and metals, which can be naturally-occurring or result from urban storm water run-off, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.

• Pesticides and herbicides, which may come from a variety of sources such as agriculture, storm water run-off, and residential uses.

• Organic chemical contaminants, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and can also come from gas stations, urban storm water runoff, and septic systems.

Radioactive contaminants, which can be naturally occurring or be the result of oil and gas production and mining activities.

In order to ensure that tap water is safe to drink, EPA prescribes regulations which limit the amount of certain contaminants in water provided by public water systems. Food and Drug Administration (FDA) regulations establish limits for contaminants in bottled water.

Some people may be more vulnerable to contaminants in drinking water than the general population. People who are immunocompromised such as cancer patients undergoing chemotherapy, organ transplant recipients, HIV/AIDS positive or other immune system disorders, some elderly, and infants can be particularly at risk from infections. People at risk should seek advice about drinking water from their health care providers. EPA/CDC guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbiological contaminants are available from the Safe Drinking Water Hotline (800-426-4791).

## Information about Lead

Lead in drinking water is rarely found in source water but is primarily from materials and components associated with service lines and home plumbing. Your water system is responsible for providing high quality drinking water but cannot control the variety of materials used in plumbing components.

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Use *only* water from the cold-water tap for drinking, cooking, and *especially for making baby formula*. Hot water is more likely to cause leaching of lead from plumbing materials. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. These recommended actions are very important to the health of your family.

Lead levels in your drinking water are likely to be higher if:

- Your home or water system has lead pipes, or
- Your home has faucets or fittings made of brass which contains some lead, or
- Your home has copper pipes with lead solder and you have naturally soft water, and
- Water often sits in the pipes for several hours.

If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water hotline or at <u>www.epa.gov/safewater/lead</u>.

## **Monitoring Information**

**Poarch Band of Creek Indians Utility Authority** routinely monitors for contaminants in your drinking water according to Federal laws, using EPA approved methods and a certified laboratory. Environmental regulations allow us to monitor for some contaminants less than once per year because the concentrations of these contaminants do not change frequently. This report contains results from the most recent monitoring which was performed in accordance with the regulatory schedule.

Constituents Monitored	Poarch Creek	Freemanville
Inorganic Contaminants	2017	2016
Lead/Copper	2015	2017
Microbiological Contaminants	current	current
Nitrates	2017	2017
Radioactive Contaminants	2015	2010
Synthetic Organic Contaminants (including pesticides and herbicides)	2017	2015
Volatile Organic Contaminants	2015	2017
Disinfection By-products	2017	2017

## **Monitoring Results**

As you can see by the following tables, our system had no violations. We have learned through our monitoring and testing that some constituents have been detected. We are pleased to report that our drinking water meets federal and state requirements. The following table shows *only* those contaminants that were detected in our water.

	POA	RCH BANI	D OF CR	EEK INC	DIANS	UTILITY AUTHORITY
	TABI	E OF DET	ECTED D	RINKIN	G WA	TER CONTAMINANTS
	Violation	Level	Unit			Likely Source
Contaminants	Y/N	Detected	Msmt	MCLG	MCL	of Contamination
Radium-226	NO	$\textbf{0.6}\pm\textbf{0.5}$	PCi/l	0	5	Erosion of natural deposits
Combined radium	NO	$3.0\pm 0.6$	PCi/l	0	5	Erosion of natural deposits
Uranium	NO	$\textbf{0.3}\pm\textbf{0.3}$	ppb	0	30	Erosion of natural deposits
Copper	NO	0.118 * 0>AL	ppm	1.3	AL=1. 3	Corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
Fluoride	NO	ND-0.61	ppm	4	4	Erosion of natural deposits; water additive which promotes strong teeth; discharge from fertilizer and chemical factories
Nitrate (as Nitrogen)	NO	0.35-1.10	ppm	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits
TTHM [Total trihalomethanes]	NO	ND-12.7	ppb	0	80	By-product of drinking water chlorination
HAA5 [Total haloacetic acids]	NO	ND-2.47	ppb	0	60	By-product of drinking water chlorination
Unregulated Contaminar						
Chloroform	NO	ND-1.44	ppb	n/a	n/a	Naturally occurring in the environment or as a result of industrial discharge or agricultural runoff
Bromodichloromethane	NO	ND-0.70	ppb	n/a	n/a	Naturally occurring in the environment or as a result of industrial discharge or agricultural runoff
Secondary Contaminant	S					
Chloride	NO	3.40-4.23	ppm	n/a	250	Naturally occurring in the environment or as a result of industrial discharge or agricultural runoff
Hardness	NO	2.03-5.83	ppm	n/a	n/a	Naturally occurring in the environment or as a result of treatment with water additives
рН	NO	7.29-7.79	S.U.	n/a	n/a	Naturally occurring in the environment or as a result of treatment with water additives
Sodium	NO	13.9-41.4	ppm	n/a	n/a	Naturally occurring in the environment
Sulfate	NO	ND-2.08	ppm	n/a	250	Naturally occurring in the environment or as a result of industrial discharge or agricultural runoff
Total Dissolved Solids	NO	28.0-136	ppm	n/a	500	Naturally occurring in the environment or as a result of industrial discharge or agricultural runoff

\* Figure shown is 90<sup>th</sup> percentile, and # of sites above (>) Action Level (AL) = 0

	FREEMANVILLE WATER SYSTEM								
	TABLE OF DETECTED DRINKING WATER CONTAMINANTS								
Contaminants	Violation Y/N	Level Detected	Unit Msmt	MCLG	MCL	Likely Source of Contamination			
Copper	NO	0.282 * 0>AL	ppm	1.3	AL=1.3	Corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives			
Nitrate (as Nitrogen)	NO	0.52-0.98	ppm	10	10	Runoff from fertilizer use; leaching from septic tanks, sewage; erosion of natural deposits			
TTHM [Total trihalomethanes]	NO	ND-3.08	ppb	0	80	By-product of drinking water chlorination			
Secondary Contaminants									
Chloride	NO	3.82-5.02	ppm	none	250	Naturally occurring in the environment or from runoff			
Hardness	NO	2.11-5.50	ppm	none	none	Naturally occurring in the environment or from treatment			
рН	NO	7.00-7.92	S.U.	none	none	Naturally occurring in the environment or from treatment			
Sodium	NO	39.3-42.5	ppm	none	none	Naturally occurring in the environment			
Sulfate	NO	0.59-0.62	ppm	none	250	Naturally occurring in the environment; erosion			
Total Dissolved Solids	NO	56.0-124	ppm	none	500	Naturally occurring in the environment or from runoff			

\* Figure shown is 90<sup>th</sup> percentile and # of sites above action level (1.3 ppm) = 0

#### Definitions

Action Level (AL)- the concentration of a contaminant that, if exceeded, triggers treatment or other requirements.

Coliform Absent (ca)-Laboratory analysis indicates that the contaminant is not present.

**Disinfection byproducts (DBPs)**- are formed when disinfectants used in water treatment plants react with bromide and/or natural organic matter (i.e., decaying vegetation) present in the source water.

Locational Running Annual Average (LRAA)-yearly average of all the DPB results at each specific sampling site in the distribution system. The highest distribution site LRAA is reported in the Table of Detected Contaminants.

**Maximum Contaminant Level (MCL)**- The MCL is the highest level of a contaminant that is allowed in drinking water. MCLs are set as close to the MCLGs as feasible using the best available treatment technology.

Maximum Contaminant Level Goa (MCLG)- The MCLG is the level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs allow for a margin of safety.

Maximum Residual Disinfectant Level (MRDL)-the highest level of a disinfectant allowed in drinking water

Millirems per year (mrem/yr)-measure of radiation absorbed by the body.

Nephelometric Turbidity Unit (NTU)-a measure of the clarity of water. Turbidity in excess of 5 NTU is just noticeable to the average person.

Non-Detects (ND)- laboratory analysis indicates that the constituent is not present above detection limits of lab equipment.

**Not Reported (NR)**-laboratory analysis, usually Secondary Contaminants, not reported by water system. EPA recommends secondary standards to water systems but does not require systems to comply.

Parts per billion (ppb) or Micrograms per liter (µg/l)-one part per billion corresponds to one minute in 2,000 years, or a single penny in \$10,000,000.

Parts per million (ppm) or Milligrams per liter (mg/l)-one part per million corresponds to one minute in two years or a single penny in \$10,000.

Parts per quadrillion (ppq) or Picograms per liter (picograms/l)-one part per quadrillion corresponds to one minute in 2,000,000,000 years, or a single penny in \$10,000,000,000,000.

Parts per trillion (ppt) or Nanograms per liter (nanograms/I)-one part per trillion corresponds to one minute in 2,000,000 years, or a single penny in \$10,000,000,000.

Picocuries per liter (pCi/L)-picocuries per liter is a measure of the radioactivity in water.

RAA–Running annual average

Standard Units (S.U.)-pH of water measures the water's balances of acids and bases and is affected by temperature and carbon dioxide gas. Water with less than 6.5 could be acidic, soft, and corrosive. A pH greater than 8.5 could indicate that the water is hard. Treatment Technique (TT)- a required process intended to reduce the level of a contaminant in drinking water.

Variances & Exemptions (V&E)-State or EPA permission not to meet an MCL or a treatment technique under certain conditions.

## **Tips on Becoming Water-Wise**

**Read Your Water Meter:** Use your water meter to check for leaks in your home. Start by turning off all faucets and water-using appliances and make sure no one uses water during the testing period. Take a reading on your water meter, wait for about 30 minutes, and then take a second reading. If the numbers have changed, you have a leak.

**Check for Leaky Toilets:** The most common source of leaks is the toilet. Check toilets for leaks by placing a few drops of food coloring in the tank. If after 15 minutes the dye shows up in the bowl, the toilet has a leak. Leaky toilets can usually be repaired inexpensively by replacing the flapper.

**Check for Leaky Faucets:** The next place to check for leaks is your sink and bathroom faucets. Dripping faucets can usually be repaired by replacing the rubber O-ring or washer inside the valve.

Following is a list of *Primary Drinking Water Contaminants* and a list of *Unregulated Contaminants* for which our water system routinely monitors. These contaminants were *not* detected in your drinking water unless they are listed in the *Table of Detected Drinking Water Contaminants*.

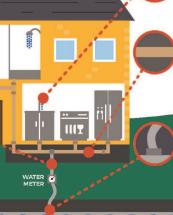
STANDAR	D LIST C	F PRIMARY DRINKING WA	ATER CONTAMINANTS		
Contaminant	MCL	Unit of Msmt	Contaminant	MCL	Unit of
Bacteriological Contaminants			trans-1,2-Dichloroethylene	100	ppb
Total Coliform Bacteria	<5%	present or absent	Dichloromethane	5	ppb
Fecal Coliform and E. coli	0	present or absent	1,2-Dichloropropane	5	ppb
Fecal Indicators (enterococci or coliphage)	0	present or absent	Di (2-ethylhexyl)adipate	400	ppb
Turbidity	TT	NTU	Di (2-ethylhexyl)phthalate	6	ppb
Cryptosporidium	TT	Calculated organisms/liter		7	ppb
Radiological Contaminants			Dioxin [2,3,7,8-TCDD]	30	ppq
Beta/photon emitters	4	mrem/yr	Diquat	20	ppb
Alpha emitters	15	pCi/l	Endothall	100	ppb
Combined radium	5	pCi/l	Endrin	2	ppb
Uranium	30	pCi/l	Epichlorohydrin	TT	TT
Inorganic Chemicals			Ethylbenzene	700	ppb
Antimony	6	ppb	Ethylene dibromide	50	ppt
Arsenic	10	ppb	Glyphosate	700	ppb
Asbestos	7	MFL	Heptachlor	400	ppt
Barium	2	ppm	Heptachlor epoxide	200	ppt
Beryllium	4	ppb	Hexachlorobenzene	1	ppb
Cadmium	5	ppb	Hexachlorocyclopentadien	50	ppb
Chromium	100	ppb	Lindane	200	ppt
Copper	AL=1.3	ppm	Methoxychlor	40	ppb
Cyanide	200	ppb	Oxamyl [Vydate]	200	ppb
Fluoride	4	ppm	Polychlorinated biphenyls	0.5	ppb
Lead	AL=15	ppb	Pentachlorophenol	1	ppb
Mercury	2	ppb	Picloram	500	ppb
Nitrate	10	ppm	Simazine	4	ppb
Nitrite	1	ppm	Styrene	100	ppb
Selenium	.05	ppm	Tetrachloroethylene	5	ppb
Thallium	.002	ppm	Toluene	1	ppm
Organic Contaminants			Toxaphene	3	ppb
2,4-D	70	ppb	2,4,5-TP(Silvex)	50	ppb
Acrylamide	TT	ŤT	1,2,4-Trichlorobenzene	.07	ppm
Alachlor	2	ppb	1,1,1-Trichloroethane	200	ppb
Benzene	5	ppb	1,1,2-Trichloroethane	5	ppb
Benzo(a)pyrene [PAHs]	200	ppt	Trichloroethylene	5	ppb
Carbofuran	40	ppb	Vinyl Chloride	2	ppb
Carbon tetrachloride	5	ppb	Xylenes	10	ppm
Chlordane	2	ppb	Disinfectants & Disinfecti	on	· · ·
Chlorobenzene	100	ppb	Chlorine	4	ppm
Dalapon	200	ppb	Chlorine Dioxide	800	ppb
Dibromochloropropane	200	ppt	Chloramines	4	ppm
o-Dichlorobenzene	600	ppb	Bromate	10	ppb
p-Dichlorobenzene	75	ррb	Chlorite	1	ppm
1,2-Dichloroethane	5	ppb	HAA5 [Total haloacetic	60	ppb
1,1-Dichloroethylene	7	ppb	TTHM [Total	80	ppb
cis-1,2-Dichloroethylene	70	ppb	-		1
		NREGULATED CONTAMIN	ANTS		
1,1 – Dichloropropene	Aldicarb		Chloroform	Metolachlor	
1,1,1,2-Tetrachloroethane		Sulfone	Chloromethane	Metribuzin	
1,1,2,2-Tetrachloroethane		Sulfoxide	Dibromochloromethane	N - Butylbenze	ene
1,1-Dichloroethane	Aldrin		Dibromomethane	Naphthalene	
1,2,3 - Trichlorobenzene	Bromobenzene		Dicamba	N-Propylbenze	ene
1,2,3 - Trichloropropane	Bromochloromethane		Dichlorodifluoromethane	O-Chlorotoluene	
1,2,4 - Trimethylbenzene	Bromochloromethane		Dieldrin	P-Chlorotoluene	
1,3 – Dichloropropane	Bromofo		Hexachlorobutadiene	P-Chlorotoluene P-Isopropyltoluene	
1,3 – Dichloropropene	Bromon		Isoprpylbenzene	Propachlor	
1,3,5 - Trimethylbenzene	Butachl		M-Dichlorobenzene	Sec - Butylber	izene
2,2 – Dichloropropane	Carbary		Methomyl	Tert - Butylber	
3-Hydroxycarbofuran	Chloroe		MTBE	Trichlorfluoron	
อากรุงกับหรือสมบานเล่า	SHIDLOF				Ictiane



## Sources of **LEAD** in Drinking Water

Copper Pipe with Lead Solder: Solder made or installed before 1986 contained high lead levels.

Lead Service Line: The service line is the pipe that runs from the water main to the home's internal plumbing. Lead service lines can be a major source of lead contamination in water.



Faucets: Fixtures inside your home may contain lead.

Galvanized Pipe: Lead particles can attach to the surface of galvanized pipes. Over time, the particles can enter your drinking water, causing elevated lead levels.

Lead Coose Necks: Goose necks and pigtails are shorter pipes that connect the lead service line to the main.

MAIN WATER LINE

## **Reduce Your Exposure To Lead**



Use only cold water for drinking, cooking and making baby formula. Boiling water does not remove lead from water. Regularly clean your faucet's screen (also known as an aerator).

Consider using a water filter certified to remove lead and know when it's time to replace the filter.



To find out for certain if you have lead in drinking water, have your water tested.

## **Replace Your Lead Service Line**



Water systems are required to replace lead service lines if a water system cannot meet EPA's Lead Action Level through optimized corrosion control treatment.

Replacement of the lead service line is often the responsibility of both the utility and homeowner. Homeowners can contact their water system to learn about how to remove the lead service line.

## **Identify Other Lead Sources In Your Home**

Lead in homes can also come from sources other than water. If you live in a home built before 1978, you may want to have your paint tested for lead. Consider contacting your doctor to have your children tested if you are concerned about lead exposure.



For more information, visit: epa.gov/safewater



PLEASE DO YOUR PART TO PROTECT YOUR FAMILY.







Time Needed

Materials List

Recycled clear plastic water bottle with cap

Blue food coloring Sharpie marker

## WATER CYCLE DISCOVERY BOTTLE

Upon completion of this activity, students will gain an understanding of how the sun provides the energy for the phase changes of the water cycle.

#### Background What is the water cycle and how

does it work? All the water that has ever existed on Earth is still here and gets cycled and recy-cled infinitely as it changes from gas to liquid to solid and back again. The sun's energy heats up water on the surface, in lakes, rivers and oceans, and causes evaporation. When this vapor hits cooler air it changes back to it's liquid form and creates clouds. This part of the water cycle is called condensation. When so much of water vapor has condensed that the droplets become large and heavy, the liquid falls back down in the form of pre-cipitation. When the water falls back down it may collect in dif-ferent bodies of water like rivers, streams, lakes or oceans. Or it may sink into the ground to feed plants, get stored in an aquifer, or it may runoff into nearby bodies of water if the ground is already saturated. Eventually the water evaporates which is all part of the water cycle and at another point it will fall back down to the ground. In the water cycle discovery bottle you can not see each stage completely, but it is a great hands-on project to go along

AQUA POWER

There are many activities related to water that can be easily orga-

nized in the after school setting that provide experiential learn-

ing in the world of engineering.

Water wheels are some of the earliest machines designed to

harness the power of water to accomplish tasks. This activity

gets kids building, thinking and experimenting. Here are a few

Teach Engineering : Water Wheel Lesson Plan <u>https://goo.</u>

Background

Upon completion of this activity, students will have used experimentation to discover the mechanics of a water wh

with talking about the water cycle with your kids. It's a simple way to provide a visual for kids to see the changes.

Here is a great resource for more information and activities about the water cycle:

#### USGS Water Science School https://goo.gl/917sae

#### Learning Target

Upon completion of this activity, students will gain an understanding of how the sun provides the energy for the phase changes of the water cycle.

## Essential QuestionsWhat are the 3 stages of the

- what are the o stages of the water cycle?Why is the sun's energy im-
- portant to the water cycle?

#### Bridge

Introduce the activity with a connection to prior knowledge... Begin by asking your students if they know of anything that "cycles" ? They might think of recycling, when an object can be used over and over again, such as a plastic bottle that can be



melted, re-molded and reused infinitely. Another great example would be the cycle of the seasons. Then ask if they are familiar with the water cycle. Most students have been exposed to this concept by mid- to late-elementary years. Review the stages of the water cycle before introducing the construction of their discovery bottles.



Engage Make it real, make it fun, help them build new knowledge...

First, have students draw a few

elements of the water cycle on

their bottles - clouds, rain drops, the sun, a lake or puddle, land. Then add about ¼ cup of water and a drop of food coloring. Seal the bottles and place them in a sunny location. Such as a window sill or perhaps outside if it is a warm season.

After about 30-60 minutes, have students observe any changes that have happened inside their bottles and discuss how these findings relate to the water cycle.

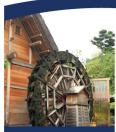
#### Enrich

Ideas for additional investigation and extension... Make a water cycle bracelet to

make a water cycle Diaceter to represent the steps or pathways of the water cycle. Use different colored beads to represent Sun, Evaporation (water vapor), Condensation (clouds), Precipitation (rain), Surface Water (lakes and ponds), Ground Water (water in the earth's aquifers and soils), Plants and Animals.

#### Reflect

What did you learn? What was surprising, fun, curious, What new question(s) do you now have?



Time Needed 40 minutes

#### Materials List

- Foam plates
   Plastic cups and spoons
   Recycled 2 liter soda
- bottles 4. Aluminum pie plates
- 5. Dowels or skewers to serve as axles
- 6. Masking or duct tape 7. String
- String
   Large plastic bin or tub for testing
   Pitcher

#### Green Kids Water Wheel Project https://goo.gl/BGZkVt

good resources:

al/baKhcS

Cup and Plate Water Wheel

#### Make it Snappy Water Wheel https://goo.gl/pDmPEL

#### Learning Target

Upon completion of this activity, students will have used experimentation to discover the mechanics of a water wheel.

### **Essential Questions**

- STEM learning experiences aren't just for the science classroom. There are many activities related • What kinds of tasks can a wa-
  - What kinds of tasks can a water wheel do?
    What water wheel designs are
  - most efficient?

#### **Bridge** Introduce the activity with a connection to prior knowledge...

Ask students to think about how water can push things, change the shape of things. Likely responses include the action of waves, the force of a hose, erosion by rivers and streams. Explain that those are all examples of the force that moving water can exert on objects. Show pictures of water wheels. Discuss the kinds of work that water wheels do for people, such as grinding flour by turning large stones and generating electricity by turning magnetic turbines.

#### Engage

Make it real, make it fun, help them build new knowledge...

In groups of 2 or 3, students will then make their own version of

a water wheel. If working with young learners, show the demonstration wheel already construct? ed that they can copy. If older? learners, show them the materials they can use to construct a water wheel and let them create the design. Test their final designs over a large plastic tote or bucket outdoors where water mess can be minimized.

#### Enrich

Ideas for additional investigation and extension...

Set up a competition to see how strong their water wheels are, by attaching a string and an object to the axle. Which water wheel can lift the heaviest object? Have student experiment with one design but different sizes to make observations about mechanical advantage. Change other variables, such as number of or angle of "paddles", cups vs. spoons as "paddles".

#### Reflect

What did you learn? What was surprising, fun, curious? What new question(s) do you have?



# **LOOKING AHEAD!**

## Escambia County Schools 2018/2019 Year Calendar

August 2018								
Su	Мо	Tu	We	Th	Fr	Sa		
			1	2	3	4		
5	6	7	8	9	10	11		
12	13	14	15	16	17	18		
19	20	21	22	23	24	25		
26	27	28	29	30	31			

September 2018								
Su	Мо	Tu	We	Th	Fr	Sa		
	_					1		
2	3	4	5	6	7	8		
9	10	11	12	13	14	15		
16	17	18	19	20	21	22		
23	24	25	26	27	28	29		
30								

	October 2018								
Su	Мо	Tu	We	Th	Fr	Sa			
	1	2	3	4	5	6			
7	8	9	10	11	12	13			
14	15	16	17	18	19	20			
21	22	23	24	25	26	27			
28	29	30	31						

	November 2018								
Su	Мо	Tu	We	Th	Fr	Sa			
				1	2	3			
4	5	6	7	8	9	10			
11	12	13	14	15	16	17			
18	19	20	21	22	23	24			
25	26	27	28	29	30				

December 2018									
Su	Мо	Tu	We	Th	Fr	Sa			
						1			
2	3	4	5	6	7	8			
9	10	11	12	13	14	15			
16	17	18	19	20	21	22			
23	24	25	26	27	28	29			
30	31								

1	80	Stu	dent	Days

## 187 Teacher Days

## Holiday (schools closed)

Sept 3 Labor Day Nov 12 Veterans' Day Nov 19-23 Thanksgiving Holidays Dec 21– Jan 4 Christmas Holidays Jan 21 Martin Luther King Day March 25-29 Spring Break April 19 Good Friday

#### Teacher Workday, No Students

Aug 3 Teacher Institute Aug 6 Professional Development Day Aug 7 Teacher Workday Oct 11 Fall Conferences (1:30-6 PM) Oct 12 Professional Development Day Feb 7 Spring Conferences (1:30-6 PM) Feb 8 Professional Development Day May 24 Teacher Workday

#### Early Release (students)

* Oct 11	* Dec 20
* Feb 7	* May 23

## End of Grading Periods

1st nine weeks:Oct 5 (42 days)2nd nine weeks:Dec 20 (47 days)3rd nine weeks:March 8 (43 days)4th nine weeks:May 23 (48 days)

### **Progress Reports**

* Sept 10	* Feb 6	
* Nov 5	* April 15	

R	eport Cards
Oct 10	1st nine weeks
Jan 9	2nd nine weeks

March 13	3rd nine weeks
May 23	4th nine weeks

## High School Graduation

May 21	Escambia County HS
May 23	W.S. Neal HS
May 24	Flomaton HS

January 2019						
Su	Мо	Tu	We	Th	Fr	Sa
		1	2	3	4	5
6	7	8	9	10	11	12
13	14	15	16	17	18	19
20	21	22	23	24	25	26
27	28	29	30	31		

February 2019						
Su	Мо	Tu	We	Th	Fr	Sa
					1	2
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28		

March 2019						
Su	Мо	Tu	We	Th	Fr	Sa
					1	2
3	4	5	6	7	8	9
10	11	12	13	14	15	16
17	18	19	20	21	22	23
24	25	26	27	28	29	30
31						

April 2019						
Su	Мо	Tu	We	Th	Fr	Sa
	1	2	3	4	5	6
7	8	9	10	11	12	13
14	15	16	17	18	19	20
21	22	23	24	25	26	27
28	29	30				

May 2019						
Su	Мо	Tu	We	Th	Fr	Sa
			1	2	3	4
5	6	7	8	9	10	11
12	13	14	15	16	17	18
19	20	21	22	23	24	25
26	27	28	29	30	31	

# **IMPORTANT NUMBERS**

## **Emergency - 911**

• After Hours Hotline for Utilities Emergencies	1-251-446-4920
• FBI (Federal Bureau of Investigation)	1-251-438-3674
National Poison Control	1-800-222-1222
• National Response Center	1-800-424-8802
Alabama Coalition Against Domestic Violence	1-800-650-6522
Alabama Forestry Commission	1-800-392-5679
Consumer Fraud	1-800-392-5658
• Adult Abuse	1-800-458-7214
Adoption & Foster Inquiry	1-866-425-5437
Child Abuse & Neglect	1-334-242-1310
National Suicide Prevention Lifeline	1-800-273-8255





Poarch Creek Indians Utilities Authority 5811 Jack Springs Road Atmore, AL 36502